

5,7-Dibromo-*N*-tosylquinolin-8-amineLuiz Everson da Silva,^{a,b}
Antonio Carlos Joussef,^a Sabine
Foro^b and Boris Schmidt^{b*}^aDepartamento de Química—UFSC, 88040-900
Florianópolis, SC, Brazil, and ^bClemens-Schöpf-
Institut für Organische Chemie und Biochemie,
Technische Universität Darmstadt,
Petersenstrasse 22, D-64287 Darmstadt,
Germany

Correspondence e-mail: foro@tu-darmstadt.de

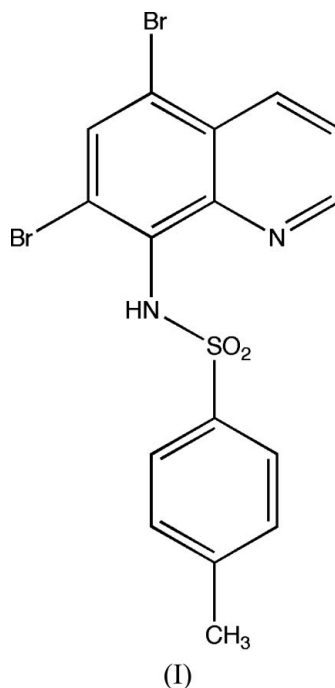
Key indicators

Single-crystal X-ray study
T = 299 K
Mean $\sigma(\text{C}-\text{C}) = 0.007 \text{ \AA}$
R factor = 0.041
wR factor = 0.109
Data-to-parameter ratio = 14.3For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.

In the crystal structure of the title compound, $\text{C}_{16}\text{H}_{12}\text{Br}_2\text{N}_2\text{O}_2\text{S}$, molecules are linked by two hydrogen bonds to form a three-dimensional network. The H atom of the NH group has two intermolecular contacts; one to a sulfonyl O atom ($\text{H}\cdots\text{O} = 2.22 \text{ \AA}$) and the other to the Br at the quinoline C-7 position ($\text{H}\cdots\text{Br} = 3.11 \text{ \AA}$).

Comment

The investigation of the crystal structure of the title compound, (I), is part of our search for Zinquin Ester analogues (Kimber *et al.*, 2000). This class of compounds has been shown to be effective in the detection of zinc (II) in a range of mammalian cells (Pearce *et al.*, 2001).



The molecular structure of the title compound (I) is illustrated in Fig. 1. The quinoline ring system is planar and inclined to the benzene ring of the tosyl moiety [torsion angle $\text{C}1-\text{N}1-\text{S}1-\text{C}10$ being $76.5(3)^\circ$].

The crystal packing of (I) is stabilized through a hydrogen-bonding network, as shown in Fig. 2. Details of the hydrogen bonding are given in Table 1.

Experimental

Compound (I) was prepared according to the literature procedure (Xue *et al.*, 2000). Suitable crystals were obtained by recrystallization from methanol-dichloromethane (1:1).

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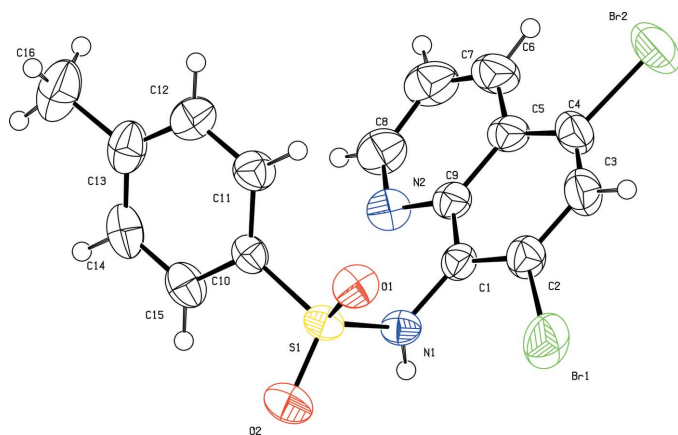


Figure 1
Molecular structure of (I), showing the atom labeling and displacement ellipsoids drawn at the 50% probability level.

Crystal data

$C_{16}H_{12}Br_2N_2O_2S$

$M_r = 456.16$

Orthorhombic, $P2_12_12_1$

$a = 4.939$ (1) Å

$b = 16.595$ (3) Å

$c = 20.254$ (4) Å

$V = 1660.1$ (6) Å³

$Z = 4$

$D_x = 1.825$ Mg m⁻³

Cu $K\alpha$ radiation

Cell parameters from 25

reflections

$\theta = 5.3$ – 18.8°

$\mu = 7.46$ mm⁻¹

$T = 299$ (2) K

Long needle, light brown

$0.65 \times 0.08 \times 0.05$ mm

Data collection

Nonius CAD-4 diffractometer

$\omega/2\theta$ scans

Absorption correction: ψ scan

(North *et al.*, 1968)

$T_{\min} = 0.486$, $T_{\max} = 0.708$

3465 measured reflections

2997 independent reflections

2758 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.019$

$\theta_{\max} = 68.0^\circ$

$h = 0 \rightarrow 5$

$k = 0 \rightarrow 19$

$l = -24 \rightarrow 24$

3 standard reflections

frequency: 120 min

intensity decay: 1.0%

Refinement

Refinement on F^2

$R[F^2 > 2\sigma(F^2)] = 0.041$

$wR(F^2) = 0.109$

$S = 1.08$

2997 reflections

209 parameters

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.076P)^2]$

$+ 0.4357P]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} < 0.001$

$\Delta\rho_{\max} = 0.69$ e Å⁻³

$\Delta\rho_{\min} = -0.92$ e Å⁻³

Extinction correction: *SHELXL97*

Extinction coefficient: 0.0012 (2)

Absolute structure: Flack (1983)

Flack parameter: -0.01 (3)

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$N1-H1N\cdots O1^i$	0.86	2.22	2.884 (4)	134
$N1-H1N\cdots Br1^i$	0.86	3.11	3.873 (3)	149

Symmetry code: (i) $x - 1, y, z$.

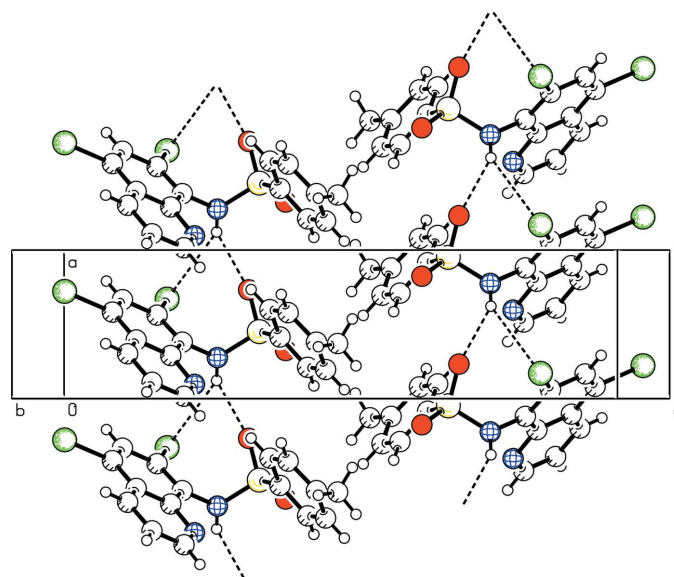


Figure 2

The packing of (I), with hydrogen bonding shown as dashed lines.

The H atoms were included in idealized positions and refined as riding atoms with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{parent atom})$, and $N-H = 0.86$ Å and $C-H = 0.93$ Å (0.96 Å for methyl groups).

Data collection: *Nonius Diffractometer Control Software* (Nonius, 1996); cell refinement: *Nonius Diffractometer Control Software*; data reduction: *REDU4* (Stoe & Cie, 1987); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *PLATON* (Spek, 2003); software used to prepare material for publication: *SHELXL97*.

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